

## WHAT IS CLAIMED:

```
W/W
                          A first glass matrix composition
                     1.
     1
          consisting essentially by mol percent of about:
                     55 < SiO_2 < 75;
                     5 < BaO < 30; and
     4
     5
                     2 < MgO < 22.
                          The first glass matrix composition of
     1
                     2.
      2
          claim 1, consisting essentially by mol percent of about:
      3
                     60 < SiO_2 < 75;
                     15 < BaO < 30; and
      4
                     7.5 < MgO < 12.5, to form a second glass matrix
      5
OOFEEHE.1EC1CO
          composition.
      6
                          A third glass matrix-ceramic particulate
      1
          composite consisting essentially by mol percent of about:
      2
                     55 < SiO_2 < 65;
      3
                     5 < BaO < 15;
      4
                     25 < MgO < 35; and
      5
                     a forsterite phase consisting of Mg<sub>2</sub>SiO<sub>4</sub>.
      6
                           The glass matrix-ceramic particulate
      1
          composite of claim 3, consisting essentially by mol
      2
          percent of about:
      3
                     57 < SiO_2 < 63
      4
                     7 < BaO < 13;
      5
                     27 < MgO < 33; and
      6
                     a forsterite phase consisting of Mg2SiO4.
      7
```

6

1

2

**4** 5

6

7

8

```
1
                    The glass matrix composition of claim 1,
               5.
2
    consisting essentially by mol percent of about:
               55 < SiO_2 < 75;
3
4
               5 < (BaO + SrO) < 30; and
5
               2 < MgO < 22.
1
                    The glass matrix-ceramic particulate
2
    composite of claim 3, consisting essentially by mol
3
    percent of about:
4
              55 < SiO_2 < 6/5;
5
               5 < (BaO + SrO) < 15; and
```

- 7. A method of making a glass matrix-ceramic particulate third composite comprising the steps of:
- (a) providing as a matrix glass, a finely divided glass powder of the glass in the composition range defined by claim 1;
- (b) providing as a particulate phase, a finely divided powder selected from the group consisting of a high expansion ceramic, a metal, and mixtures thereof;
- 9 (c) intermixing the matrix glass with the
- 10 particulate phase in an organic vehicle; and
- 11 (d) firing the intermixed materials to a
- 12 sealing temperature from 1/100 to 1250°C.

25 < MgO < 35.

- 1 8. The method of claim 7, wherein the 2 particulate phase comprises a ceramic particulate.
- 1 9. The method of claim 8, wherein the
- 2 ceramic particulate comprises a forsterite phase
- 3 consisting of Mg<sub>2</sub>SiO<sub>4</sub>.



1 10. The method of claim 7, wherein the step of providing a particulate phase comprises the step of providing a finely divided powder of a high expansion metal to form an interconnecting and current collecting material.

1 11. The method of claim 10, wherein the step 2 of providing a finely divided powder comprises providing 3 silver.

1 12. The method of claim 10, wherein the step 2 of providing a finely divided powder comprises providing 3 ferritic stainless steel.

Color